

ATTACHMENT 5.

Kingdom of Saudi Arabia

The National Commission for Academic Accreditation & Assessment

T6. Course Specifications (CS)

Advanced Analytical Chemistry

(402711-2)





Course Specifications

Institution: Umm Al-qura University	Date: 2017		
College/Department: Faculty of Applied Science / Department of Chemistry			
A. Course Identification and General Information			
1. Course title and code: Advanced Analytical Chemistry / 402711–2			
2. Credit hours: 2 hrs.(Theoretical)			
3. Program(s) in which the course is of	fered: Ph. D. in Chemistry		
4. Name of faculty member responsible	e for the course: Dr. Amr Lotfy Saber		
5. Level/year at which this course is of	fered: 1 st / 1 st		
6. Pre-requisites for this course (if any)	: not applicable		
7. Co-requisites for this course (if any)	: not applicable		
8. Location if not on main campus: El-	Abedyah, El-Azizya, and El-Zaher		
9. Mode of Instruction (mark all that ap	oply)		
a. traditional classroom	What percentage?		
b. blended (traditional and online)	What percentage? 100%		
c. e-learning	What percentage?		
d. correspondence	What percentage?		
f. other	What percentage?		
Comments:			



B Objectives

1. What is the main purpose for this course?

By completing this course, the students will be able to:

- Explain the principles of hyphenated chromatographic techniques.
- Compare between different detectors which are used with liquid chromatographic techniques.
- Classify the different types of hyphenated chromatography.
- Use hyphenated chromatographic techniques for the analysis of different samples.
- Compare between the advantages of deferent detectors coupling with liquid chromatography.
- 2. Briefly describe any plans for developing and improving the course that are being implemented. (e.g. increased use of IT or web based reference material, changes in content as a result of new research in the field)
- Changes in content as a result of new research in the field.
- Encourage students to carry out research reports related to the course subjects using the library, data base services, and/or websites.
- Increased use of IT or web based reference material.
- The use of smart teaching halls for lectures.
 - C. Course Description (Note: General description in the form used in Bulletin or handbook)

Course Description:

	No. of	Contact
OURA UNIVERSITY	Weeks	hours
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To anset the	2	4
	DE CHARACTER OF CH	Weeks 2



1	2
1	2
1	2
1	2
1	2
1	2
2	4
1	2
	1 1 1

2. Course components (total contact hours and credits per semester):						
	Lecture	Tutorial	Laboratory or Studio	Practical	Other:	Total
Contact Hours	26	-		-		26
Credit	2	-		-		2

3. Additional private study/learning hours expected for students per week.	2 hrs	

4. Course Learning Outcomes in NQF Domains of Learning and Alignment with Assessment Methods and Teaching Strategy

On the table below are the five NQF Learning Domains, numbered in the left column.

<u>First</u>, insert the suitable and measurable course learning outcomes required in the appropriate learning domains (see suggestions below the table). <u>Second</u>, insert supporting teaching strategies that fit and align with the assessment methods and intended learning outcomes. <u>Third</u>, insert appropriate assessment methods that accurately measure and evaluate the learning outcome. Each course learning



outcomes, assessment method, and teaching strategy ought to reasonably fit and flow together as an integrated learning and teaching process. (Courses are not required to include learning outcomes from each domain.)

Code	NQF Learning Domains	Course Teaching	Course
#	And Course Learning Outcomes	Strategies	Assessment
			Methods
1.0	Knowledge		
1.1	Describe the analytical aspects of hyphenated	• Lectures	•Exams
	chromatographic techniques	Scientific discussion	• web-based
1.2	Select the procedures for coupling between	• Library visits	student
	mass spectrometry and separation science	Web-based study	performance
1.3	Determine the relation between the	• Using open	systems
	evaporative light scattering and charged	discussion to link the	• portfolios
	aerosol detector	previous knowledge	•long and short
1.4	Familiar with coupling liquid chromatography	to the current and	essays
	and other separation techniques to NMRS	future topics	
1.5	Know information about liquid-phase	• The students use the	
	chemiluminescence detection for HPLC	internet to prepare an	
1.6	Recognize the multidimensional High-	essay about a recent	
	Performance Liquid Chromatography	advances related to	
1.7	Explain the application of infrared and raman	the course	
	spectroscopy for detection in liquid		
	chromatographic separations		
2.0	Cognitive Skills		
2.1	Construct the coupling between mass	• Lectures	• Exams
	spectrometry and separation science.		



2.2	Discover the application of Infrared and	Scientific discussion	• web-based
	Raman Spectroscopy for detection in liquid	• Library visits	student
	chromatographic separations.	Web-based study	performance
2.3	Formulate the information about liquid-phase	• Using brain storming	systems
	chemiluminescence detection for HPLC	at the beginning of	• portfolios
2.4	Develop the coupling of liquid	each lecture in order	•long and short
	chromatography and other separation	to stimulate the	essays
	techniques to NMRS.	students towards the	• Through
2.5	Report the relation between the evaporative	new topic of the	assignments
	light scattering and charged aerosol detector.	course.	and homework.
		• Enhancing open	
		discussion during the	
		lecture.	
3.0	Interpersonal Skills & Responsibility		
3.0 3.1	Interpersonal Skills & Responsibility Take the personality and responsibility for	•Encourage the	• Homework and
		•Encourage the solving problems in	Homework and group reports
	Take the personality and responsibility for		
3.1	Take the personality and responsibility for their own learning.	solving problems in	
3.1	Take the personality and responsibility for their own learning. Work effectively in groups and exercise	solving problems in groups during	
3.1	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate.	solving problems in groups during lecture.	
3.1	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate. Act ethically and consistently with high molar	solving problems in groups during lecture. • Making open	
3.1	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate. Act ethically and consistently with high molar standards in personal and public forums.	solving problems in groups during lecture. • Making open discussion about	
3.1	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate. Act ethically and consistently with high molar standards in personal and public forums.	solving problems in groups during lecture. • Making open discussion about certain recent topic of the course.	
3.1 3.2 3.3	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate. Act ethically and consistently with high molar standards in personal and public forums. Community linked thinking.	solving problems in groups during lecture. • Making open discussion about certain recent topic of the course.	
3.1 3.2 3.3 4.0	Take the personality and responsibility for their own learning. Work effectively in groups and exercise leadership when appropriate. Act ethically and consistently with high molar standards in personal and public forums. Community linked thinking.	solving problems in groups during lecture. • Making open discussion about certain recent topic of the course. umerical	group reports



4.2	Use information and communication	• Scientific discussion	• web-based
	technologies	• Library visits	student
4.3	Use basic mathematical and statistical	• Web-based study	performance
	techniques.		systems
			• portfolios
			•long and short
			essays
5.0	Psychomotor: NOT APPLICABLE	1	I

5. Schedule of Assessment Tasks for Students During the Semester			
	Assessment task (e.g. essay, test, group project, examination, speech, oral presentation, etc.)	Week Due	Proportion of Total Assessment
1	Activities and Assignments.		10 %
2	Midterm Exam.	8	30 %
3	Final Exam.	15-16	60 %
4	Total	100 %	

D. Student Academic Counseling and Support

- 1. Arrangements for availability of faculty and teaching staff for individual student consultations and academic advice. (include amount of time teaching staff are expected to be available each week)
 - We have faculty members to provide counselling and advice.
 - Office hours: During the working hours weekly.
 - Academic Advising for students.

E Learning Resources

- 1. List Required Textbooks
- -R. Andrew Shalliker "Hyphenated and Alternative Methods of Detection in Chromatography" CRC Taylor & Francis Group, LLC, Copyright © 2012.



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-David Harvey "Modern Analytical Chemistry" Copyright © 2000. Exclusive rights by The McGraw-Hill Companies, Inc. for manufacture and export INTERNATIONAL EDITION ISBN 0–07–116953–9.

- 2. List Essential References Materials (Journals, Reports, etc.)
- Lecture Handouts available on the coordinator website
- 3. List Recommended Textbooks and Reference Material (Journals, Reports, etc)
- R. Andrew Shalliker "Hyphenated and Alternative Methods of Detection in Chromatography" CRC Taylor & Francis Group, LLC, Copyright © 2012.
- -Gary D. Christian, Purnendu K. Dasgupta and Kevin A. Schug, Analytical Chemistry, 7th edition, WILEY (2014).
- 4. List Electronic Materials, Web Sites, Facebook, Twitter, etc.
 - http://www.chemweb.com
 - http://www.sciencedirect.com
 - http://www.rsc.org
- 5. Other learning material such as computer-based programs/CD, professional standards or regulations and software.

F. Facilities Required

Indicate requirements for the course including size of classrooms and laboratories (i.e. number of seats in classrooms and laboratories, extent of computer access etc.)

- 1. Accommodation (Classrooms, laboratories, demonstration rooms/labs, etc.)
 - Equippedclassrooms.
 - Providing hall of teaching aids including computers and projector.
- 2. Computing resources (AV, data show, Smart Board, software, etc.)
 - Room equipped with computer and projector and TV.
- 3. Other resources (specify, e.g. if specific laboratory equipment is required, list requirements or attach

list): No other requirements

G Course Evaluation and Improvement Processes



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- 1 Strategies for Obtaining Student Feedback on Effectiveness of Teaching
 - Complete the questionnaire evaluation of the course in particular
- 2 Other Strategies for Evaluation of Teaching by the Instructor or by the Department
 - Observations and the assistance of colleagues.
 - Independent evaluation for extent to achieve students the standards.
 - Independent advice of the duties and tasks.
- 3 Processes for Improvement of Teaching
 - Workshops for teaching methods.
 - Continuous training of member staff.
 - The application of e-learning.
 - Exchange of experiences internal and external.
- 4. Processes for Verifying Standards of Student Achievement (e.g. check marking by an independent member teaching staff of a sample of student work, periodic exchange and remarking of tests or a sample of assignments with staff at another institution)
 - Check marking of a sample of exam papers, or student work.
 - Exchange corrected sample of assignments or exam basis with another staff member for the same course in other faculty.
- 5 Describe the planning arrangements for periodically reviewing course effectiveness and planning for improvement.
 - Periodic review of the contents of the syllabus and modify the negatives.
 - Hosting a visiting staff to evaluate of the course.
 - Workshops for teachers of the course.

Nome of Instructory Dr. Amer I offer Cohon

Name of instructor. Dr. Amr Lotty Sabe	Γ
Signature:	_Date Report Completed:_10/1/2017_
Name of Field Experience Teaching Staff	_Analytical Chemistry
Program Coordinator:	
Signature:	Date Received: